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particular ompound is  
composed of elements  
A and B, the ratio of  
the mass of B to the  
mass of A will alw ys  
be the same. This is a  
statement of the law of  
exactly 12 g of  
carbon-12 is referred  
to as a(n) 11.San  
Ramon Valley High  
SchoolCHAPTER 3  
REVIEW Atoms: The  
Building Blocks of  
Matter SECTION 3  
SHORT ANSWER  
Answer the following  
questions in the space  
provided. 1. Explain  
the difference between  
the mass numberand  
the atomic numberof a  
nuclide. Mass number  
is the total number of

protons and neutrons in the nucleus of an isotope.3 Atoms: The Building Blocks of MatterHolt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.Modern Chemistry Chapter 3 Test B AnswersCH 3 Pre Test . CH 3 Nuclear Equations . CH 3 Nuclear Equations II . CH 3 Balancing Nuclear Equations. CH 3 Nuclear Chemistry . CH 3 Pre Test Calculations. CH 3 Protons-Neutrons-Electrons-Mass #-

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the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

3 Atoms: The Building Blocks of Matter

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Chapter Test A, continued Use this figure to answer questions 7 and 8.

\_\_\_\_\_ 7. A solution containing 35 g of Li<sub>2</sub>SO<sub>4</sub> dissolved in 100 g of water is heated from 10°C to 90°C. According to information in the figure, this temperature change would result in a. an additional 5 g of Li<sub>2</sub>SO<sub>4</sub> in ...

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CHOICE I. In the modern periodic table, elements are arranged according to

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Atoms: The Building Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1.

Explain the difference between the mass number and the atomic number of a nuclide.

Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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\_\_\_\_\_ 7. A solution  
containing 35 g of Li  
2SO 4 dissolved in 100  
g of water is heated  
from 10°C to 90°C.

According to  
information in the  
figure, this  
temperature change  
would result in a. an  
additional 5 g of Li 2SO  
4 in ...

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